

1. Draw Lewis structures for the following **covalent** compounds: *Show all lone pairs!*

H ₂ v.e. =	F ₂ v.e. =	O ₂ v.e. =
N ₂ v.e. =	HCl v.e. =	HBr v.e. =
H ₂ O v.e. =	H ₂ S v.e. =	NH ₃ v.e. =
CH ₄ v.e. =	CCl ₄ v.e. =	NCl ₃ v.e. =
CS ₂ v.e. =	CHCl ₃ v.e. =	HCN v.e. =